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A huff-and-puff production of gas hydrate deposits in Shenhu area of South China Sea through a vertical well

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ABSTRACT

The Shenhu area on the northern continental slope of the South China Sea is one of the most promising fields for gas hydrate exploitation. Drilling and sampling has indicated high saturations of methane hydrate in clay silty sediments at drilling site SH2. The hydrate-bearing layer is overlain and underlain by permeable zones of mobile water, and the system does not appear to be bounded by low-permeability strata. In this study a huffand-puff method is used to producing gas from the hydrate accumulation. We simulate numerically the hydrate dissociation and gas production by alternately injecting hot water and producing fluids at a vertical well. The simulations show the gas production rate in huff-and-puff operations is very small (50-140 m³/ d), and unacceptable for commercial production. The calculation also indicated secondary hydrates forms at the very early period of injecting operations, and then gas is released due to the thermal stimulation of hot water, but the amount of released gas in the injection periods of hot water is much smaller than that converts into secondary hydrates. In the production operations, much of gas is released from the hydrates due to a small depressurization at the well, but the released gas can not produce effectively due to the small pressure gradient, and thus remains in the reservoir and converts into secondary hydrates in the next operation of injecting hot water. The study provides an insight into the production potential of the Shenhu hydrate accumulations through the huff-and-puff method, and a basis for the analysis of the economic feasibility of gas production from that area.

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1. Introduction

Gas hydrates are crystalline substances composed of water and gas, in which a solid water lattice accommodates gas molecules in a cagelike structure (Sloan, 1998). The estimated amount of gas in the hydrate accumulations of the world greatly exceeds the volume of known conventional gas resources (Sloan, 1998). Because of their potential importance as an energy resource, CH₄-hydrates are currently attracting significant attention. However, the role that gas hydrates may play in contributing to the world's energy requirements will depend ultimately on the availability of producible gas hydrate resources and the cost to extract them (Moridis et al., 2004).

The four main methods of hydrate dissociation for producing gas include (Makogon, 1987, 1997): (1) depressurization, in which the pressure is lowered to a level lower than the hydration pressure P_H at the prevailing temperature (Moridis et al., 2007, 2009a); (2) thermal

stimulation, in which the temperature is raised above the hydration temperature T_H at the prevailing pressure (Moridis et al., 2004); (3) the use of inhibitors (such as salts and alcohols), which causes a shift in the P_H-T_H equilibrium through competition with the hydrate for guest and host molecules (Sloan, 1998); and (4) a combination of these methods (Li et al., 2010). In gas production from hydrate deposits depressurization and thermal stimulation are considered as the relatively feasible method both economically and technically (Moridis and Reagan, 2007a, 2007b; Moridis et al., 2009a, 2009b, 2011). The Mallik 2002 well demonstrated proof of concept that it is possible to recover energy from permafrost hydrates combining dissociation techniques of depressurization and thermal stimulation (Sloan and Koh, 2008). The huff-and-puff method, also known as cyclic steam stimulation (CSS), was accidentally discovered by Shell Oil Company in 1960 during a Venezuela recovery project, and is widely used in the oil industry to enhance oil recovery (Sayegh and Maini, 1984; Vittoratos, 1991; Leaute and Carey, 2007). The hot water, hot brine or steam huff-and-puff method is a special form of the combination of depressurization and thermal stimulation methods for gas production from hydrate deposit (Li et al., 2011).

Shenhu is near southeast of Shenhu Underwater Sandy Bench in the middle of the north slope of the SCS, between Xisha Trough and

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Dongsha Islands (Fig. 1). Geological, geophysical, geothermal, and geochemical investigations have suggested that Shenhu area is a favorable place for the formation of natural gas hydrates (Wu et al., 2008, 2010). Based on the indications of hydrate presence, five sites were selected for deep drilling and sampling in the Shenhu area (Fig. 1), and core samples were collected chronologically from SH3, SH1, SH2, SH7, and SH5 (Wu et al., 2008). Depressurization experiments verified the existence of methane gas hydrates in the clay silty cores sampled from SH2, SH3, and SH7 at the water depth of 1108–1235 m. (Li et al., 2010; Su et al., 2011). The Hydrate-Bearing Layer (hereafter referred to as HBL) is 40 m thick and the hydrate saturation ranges from 0 to 48% at drilling site SH2 which implies a huge amount of natural gas stored in the hydrate deposits. Some simulations have provided the first insight into the technical feasibility of gas production from the promising accumulations by depressurization method (Li et al., 2010; Su et al., 2011).

The main objective of this study is to assess the production potential of the hydrate deposit at the drilling Site SH2 by means of huff-and-puff. Parameters used in the reference cases are measured from the samples, such as saturations, temperature and pressure, and deposit permeability is estimated based on the properties lithology. Because of the possible imprecision of the measured data from the unconsolidated samples and uncertain geological structure, a large number of these parameters are treated as perturbation variables in the ensuing sensitivity analysis.

In evaluating the production potential of hydrate deposits in huffand-puff operations, we use the two criteria, an absolute criterion and a relative criterion (Moridis et al., 2009a). To satisfy the absolute criterion, a large production potential must be demonstrated, as quantified by a large gas production rate Q_P , a large cumulative gas production volume V_P over the duration of the production. The relative criterion is satisfied when the gas-to-water ratio $R_{GW} = V_G/V_W$ is high, indicating more gas produced relative to water production.

2. Production design

In this study we focused on the hydrate deposits of Classes 2 that occurs at the site SH2, which is also one of the most common class of hydrate accumulations in both the permafrost and in the oceans (Moridis and Reagan, 2007a; Moridis et al., 2009a). Note that the reference case of Class 2 accumulation is confined between a permeable overburden and a permeable underburden. Without confining boundaries of impermeable strata, gas production can be disappointing because flow through the boundaries limits the effectiveness of depressurization and leads to large production volumes of undesirable water (Moridis and Kowalsky, 2006; Moridis et al., 2009a). Lack of a confining overburden could lead to gas loss though the overburden toward the surface (Moridis et al., 2009a).

1129

114

110°

A same configuration of well system is used in this study for the gas production from the SH3 hydrate deposit. It is a modification of the well design used in the previous hydrate production research (Moridis et al., 2009a). The perforated interval covers the mid section (14 m) of the well in the HBL that is overlain and underlain by the permeable WZs (Fig. 2). This design has significant advantages: the configuration impedes water in the WZs from flow directly into the well in the early stage of the production, thus may effectively reduce the water production as well as the production cost. This design can cause significant dissociation around the well and yield the largest production rates in a shot run, but when the hydrate layer around the well is broken through, it may result in a sharp drop in rates of hydrate dissociation and gas production, and more water produced.

An earlier study appears to indicate that depressurization is the most promising dissociation method in the majority of hydrate deposits because of its simplicity, its technical and economic effectiveness, the fast response of hydrates to the rapidly propagating pressure wave, the near-incompressibility of water, and the large heat capacity of water (Moridis and Sloan, 2007; Moridis et al., 2007). The other dissociation methods can enhance gas production from hydrate deposits when they are used in conjunction with depressurization (Li et al., 2010), but it is not clear that a strategy using a huff-and-puff method at a vertical well in Shenhu hydrate production. In this study, a huff-and-puff method is used to produce gas from the hydrate deposits in the Shenhu area and discussed in the following part.

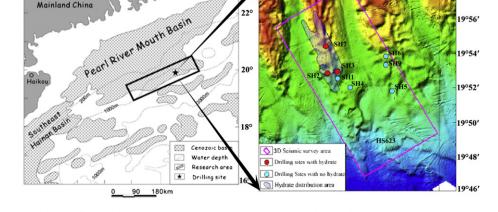
3. Numerical models and simulation approach

115°10' 115°12' 115°14'

The numerical studies in this paper were conducted using the TOUGH + HYDRATE simulator (Moridis et al., 2009b). This code can model the non-isothermal hydration reaction, phase behavior, and flow of fluids and heat under conditions typical of natural CH4-hydrate deposits in complex geologic media. It includes both an equilibrium and a kinetic model (Kim et al., 1987; Clarke and Bishnoi, 2000) of hydrate formation and dissociation. The model accounts for heat and up to four mass components (i.e., water, CH₄, hydrate, and water-soluble inhibitors such as salts or alcohols) that are partitioned among four possible phases: gas, aqueous liquid, ice, and hydrate. A total of 15 states (phase combinations) can be described by the code, which can handle any combination of hydrate dissociation mechanisms and can describe the phase changes and steep solution surfaces that are typical of hydrate problems (Moridis et al., 2009b).

The geologic system in this study corresponds to a location at the drilling site of SH2 in Shenhu area where the sea floor is at an elevation of z = -1235 m. The HBL is 40 m thick, and overlain by a permeable overburden of 188 m thick and underlain by a permeable zone.

115°16′



118°

115°6'

115°8'

1169

Fig. 1. Location of research field and drilling site in Shenhu area on northern continental slope of the South China Sea (Wu et al., 2008, 2010).

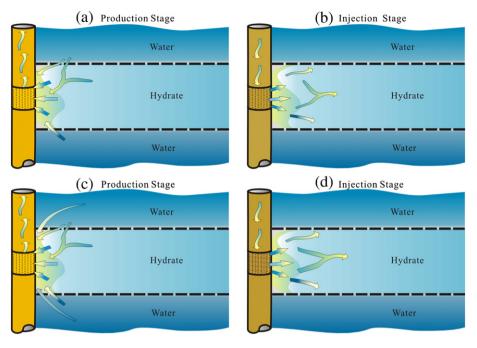


Fig.2. Well design used in the gas production from Shenhu hydrate deposits in the Southern China Sea (modified from Moridis et al., 2009a). The screened interval is used to inject hot water in the huff-and-puff operations.

Both the overburden and underburden are unbounded and typical of water zones (WZs).

The geometry and configuration of the Class 2 system at the drilling site of SH2 are shown in Fig. 3. A 20 m overburden was considered in the simulations to allow heat exchange with the laminar hydrate deposit during a 3-yr long production period. Similarly, a 20 m underburden was sufficient to provide accurate estimates of heat transfer compared to that of gas production from hydrate accumulation in the Ulleung Basin of the Korean East Sea (Moridis et al., 2009a). The acceptability

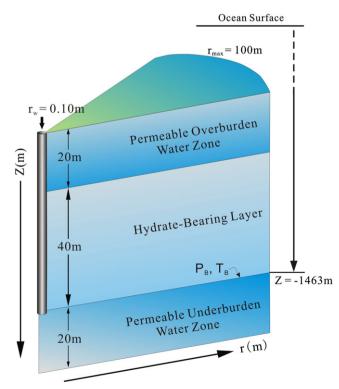


Fig.3. A schematic of the Shenhu hydrate deposits simulated in this study (modified from Moridis et al., 2009a).

is approved in the simulations. The well at the center of this cylindrical hydrate deposit had a radius $r_w = 0.1$ m. A no-flow boundary (of fluids and heat) was applied at the reservoir at radius $r_{max} = 100$ m, we also think the scale is enough for thermal transfer and fluid flow in production duration of 3 years by learning efficiency of heat transfer and previous research (Li et al., 2010; Su et al., 2011).

The same grid and media properties were used in the simulations. The cylindrical domain was discretized into $105 \times 242 = 25,410$ gridblocks in (*r*, *z*), of which 25,200 were active (the remaining being boundary cells). The uppermost and lowermost layers corresponded to constant *T*. Because the vicinity of the wellbore had been shown to be critically important to production (Moridis et al., 2009a), a very fine discretization was used, and the interval increased exponentially along the r direction. The HBL was subdivided into segments of $\Delta z = 0.25$ m each along the *z*-direction for accurate predictions, but a coarser discretization along the z axis is permissible in the WZ (Moridis et al., 2009a). We also assumed the hydrate dissociation is an equilibrium reaction (Kim et al., 1987; Moridis et al., 2009a). The well design and important parameters were referenced from the early research (Moridis et al., 2009a).

The initial conditions in the reservoir were determined by following the initialization process described by Moridis et al. (2007, 2009a). Knowing (a) the elevation at the base of the HBL, and (b) some temperatures (Table 1), we determined the local geothermal gradient G_T as 0.047 °C/m, and delineated the temperature profile at site SH2. Then the pressure profile was obtained by means of a short simulation with the temperature, salinity, and hydrate saturation.

4. A huff-and-puff production

Thermal stimulation is considered as a complementary way for dissociating hydrate by depressurization due to the limitation of thermal conduction in porous sediment (Moridis et al., 2009a; Su et al., 2011). Heat conducts along the *z*-axis direction from the heated well but fluids (water and gas) flow inversely. There is hereby a possibility that heat may be taken back by fluids flowing towards the well before it reaches the hydrate dissociation front, and thus the heat affecting range will be greatly undermined (Su et al., 2011). The former research shows that the heat affecting range is generally are very small and most heat is

Table 1

Reference hydrate deposit properties and parameters in simulations.

ParameterValueWater zone (WZ) thickness (overburden & underburden)20 mHydrate zone (HBL) thickness40 mInitial pressure P_B (at base of HBL) $1.497 \times 10^7 Pa$ Initial temperature T_B (at base of HBL) $1.497 \times 10^7 Pa$ Initial temperature T_B (at base of HBL) $1.497 \times 10^7 Pa$ Initial temperature T_B (at base of HBL) $1.497 \times 10^7 Pa$ Initial saturations in the HBL $S_H = 0.40, S_A = 0.60$ Water salinity (mass fraction) 0.030 Intrinsic permeability $k_r = k_z$ (HBL $1.0 \times 10^{-14} m^2 (=10 mD)$ and WZ)Grain density ρ_R (HBL and WZ)Porosity φ (HBL and WZ) 0.38 Porosity φ (HBL and WZ) 0.38 Hydraulic diffusion k_g (Cathles, 2007) $k_g = k/\varphi C_g M_g$ Compressibility C_g $1.0 W/m/K$ (all formations) $1.0 W/m/K$ Wet thermal conductivity k_{0RD} $1.0 W/m/K$ (all formations) $8_{0C} = k_{0RD} + (S_A^{1/2} + S_H^{1/2}) (k_{0RW} - k_{0RD}) + \varphi S_A^{1/2} + S_{1/2}^{1/2}) (k_{0RW} - k_{0RD}) + \varphi S_A^{1/2} + S_{1/2}^{1/2}) (k_{0RW} - k_{0RD}) + \varphi S_A^{1/2} + S_{1/2}^{1/2}) (k_{0RW} - k_{0RD}) + (S_A^{1/2} + $		
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(all formations)3.1 W/m/K(all formations)3.1 W/m/K(all formations)3.1 W/m/K(all formations)composite thermal conductivity model(Moridis et al., 2007) $k_{\Theta CD} + (S_{1}^{1/2} + S_{H}^{1/2}) (k_{\Theta RW} - k_{\Theta RD}) + \varphi S_{1}k_{\Theta I}$ Capillary pressure model (van Genuchten, 1980) $P_{cap} = -P_0[(S^*)^{-1/\lambda} - 1]^{-\lambda} S^* = (S_A - S_{irA})/(S_{mxA} - S_{irA})$ S _{irA} 1 λ 0.45P_010 ⁵ PaRelative permeability model (Moridis et al., 2008) $k_{rG} = (S_A^*)^n$ $s_G^* = (S_A - S_{irA})/(1 - S_{irA})$ $S_G^* = (S_A - S_{irA})/(1 - S_{irA})$ S_{irG} 0.03	Compressibility C_g	
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$ \begin{array}{ll} \mbox{Composite thermal conductivity model} & k_{\theta C} = k_{\theta RD} + (S_A^{1/2} + S_H^{1/2}) (k_{\theta RW} - k_{\theta RD}) + \varphi S_I k_{\theta I} \\ \mbox{Capillary pressure model (van Genuchten, 1980)} & P_{cap} = -P_0 [(S^*)^{-1/\lambda} - 1]^{-\lambda} S^* = (S_A - S_{IrA}) / (S_{mxA} - S_{IrA}) \\ \mbox{A} & 0.45 \\ P_0 & 10^5 \mbox{Pa} \\ \mbox{Relative permeability model (Moridis et al., 2008)} & k_{rG} = (S_A^*)^n \\ \mbox{A} & K_{rG} = (S_G^*)^n \\ \mbox{S}_{irG}^* = (S_A - S_{IrA}) / (1 - S_{IrA}) \\ \mbox{OPM model} & 5 \\ \mbox{0.03} \end{array} $		3.1 W/m/K
(Moridis et al., 2007) $k_{ORD} + \varphi S_I k_{OI}$ $h \to V \text{ com}$ Capillary pressure model (van Genuchten, 1980) $P_{cap} = -P_0[(S^*)^{-1/\lambda} - 1]^{-\lambda} S^* =$ $(S_A - S_{IrA})/(S_{mxA} - S_{IrA})$ S_{IrA} 1 λ 0.45 P_0 10 ⁵ Pa $Relative permeability model (Moridis etal., 2008)k_{rG} = (S_G^*)^nS_A^* = (S_A - S_{IrA})/(1 - S_{IrA})S_G^* = (S_G - S_{IrG})/(1 - S_{IrA})OPM modeln (Moridis et al., 2008)5S_{IrG}0.03$		
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		$k_{\Theta RD}$) + $\varphi S_I k_{\Theta I}$
$S_{irA} = (S_{r}^{A})^{n}$ $\lambda = (S_{r}^{A})^{n}$ $Relative permeability model (Moridis et al., 2008)$ $k_{rG} = (S_{r}^{A})^{n}$		
$\begin{array}{ccc} \lambda & 0.45 \\ P_0 & 10^5 \mathrm{Pa} \\ \text{Relative permeability model (Moridis et al., 2008)} & k_{rG} = (S_A^*)^n \\ & k_{rG} = (S_A^*)^n \\ & S_A^* = (S_A - S_{irA})/(1 - S_{irA}) \\ & S_C^* = (S_G - S_{irG})/(1 - S_{irA}) \\ & OPM \ \text{model} \\ & f \\ S_{irG} & 0.03 \end{array}$		(// //// //// ////
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Relative permeability model (Moridis et al., 2008) $k_{rA} = (S_A^*)^n$ $k_{rG} = (S_G^*)^n$ $S_A^* = (S_A - S_{irA})/(1 - S_{irA})$ $S_G^* = (S_G - S_{irG})/(1 - S_{irA})$ OPM model S_{irG} n (Moridis et al., 2008)5 0.03		
al., 2008) $k_{rG} = (S_G^*)^n$ $S_A^* = (S_A - S_{irA})/(1 - S_{irA})$ $S_G^* = (S_G - S_{irG})/(1 - S_{irA})$ OPM model n (Moridis et al., 2008) S_{irG} 0.03	5	
$\begin{array}{c} S_A^{\#} = (S_A - S_{irA})/(1 - S_{irA}) \\ S_G^{\#} = (S_G - S_{irG})/(1 - S_{irA}) \\ OPM \ model \\ n \ (Moridis \ et \ al., 2008) \\ S_{irG} \\ \end{array}$		
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n (Moridis et al., 2008) 5 S _{irG} 0.03		$S_G^* = (S_G - S_{irG})/(1 - S_{irA})$
<i>S</i> _{<i>irG</i>} 0.03		OPM model
10	n (Moridis et al., 2008)	5
S _{irA} 0.30	S _{irG}	
	S _{irA}	0.30

wasted. Therefore this simultaneous heating and producing way is unfeasible practically for producing gas from hydrate deposits (Su et al., 2011). It may be worse and technically unpractical if the hydrate dissociation is induced by continuously injecting hot water at a single well. The bottom-hole pressure should be slighter lower than that in reservoir in order to produce gas, which just violates the need for injecting water. But it may be viable to dissociate hydrate by alternately producing fluids and injecting hot water (referred to as huff-and-puff method).

Huff-and-puff widely used in petroleum industry is designed to produce gas from the hydrate deposits by combining mass production and hot water injection. The operations of production and injection are circulatory through the same screened interval of the well shown in Fig. 2. Production stops for injecting hot water, and then injection stops for producing fluids. In each circle the cumulative mass produced should be higher than the cumulative mass of water injected. The rate for water injection should be very small due to the low intrinsic permeability of the HBL at Shenhu area. High injection rate may incur secondary hydrate formation at the front of hydrate dissociation and even sediment failure. Of course, the rate for mass production should be strictly controlled according to the water injection rate. High production rate may incur negative pressure in the production well and excessive water as byproduct. Herewith, we recommend the rate values should obey a relation of $1.2 \times Q_{Inj} \frac{t_{Inj}}{t_{Inj}} \leq Q_{Pro} \leq 1.8 \times Q_{Inj} \frac{t_{Inj}}{t_{Inj}} (Q_{Pro} \text{ is production})$ rate and Q_{Inj} is injection rate).

4.1. Production scheme of $1_P 2_I$

Fluids production and water injection are operated alternately to get gas produced and hydrate dissociated by heat stimulation. Figs. 4 and 5 show performance of the hydrate deposits dissociated by huff-and-puff operations of 1-day production and 2-day injection (1_P2_1) in the first tentative study. Water of 80 °C is injected at a

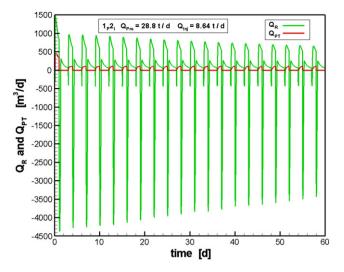


Fig.4. Evolution of gas release rate from hydrate deposits Q_R and gas production rate Q_{PT} by huff-and-puff of $1_P 2_1$.

constant rate of 0.033 kg/s (=8.64 t/d) while the fluids (water plus gas) are produced at a constant mass rate of 0.1 kg/s (=28.8 t/d). To get enough porous space for water injection, we start this operation by 1-day producing, and then convert to 2-day injection in the simulation. The zigzag type of Q_R in Fig. 4 and V_R in Fig. 5 indicate the alternate production and injection that resulting in gas releasing and secondary hydrate formation.

In the first time period (3 days) of the huff-and-puff operation both gas release and gas production are intensive. The volume of gas produced at the well ΔV_{PT}^{1} is 400 m³ (pointed by the first and highest gas producing rate in Fig. 4 and the first jump of V_{PT} in Fig. 5) but the volume of gas released from hydrate $\Delta V_{R'}^{1}$ is 1196 m³ (shown by the maximum peak in Fig. 4 and the first peak of V_R in Fig. 5) within the same production interval (1 day), which suggests that the hydrate dissociation is induced by depressurization during the production process and that the released gas can not be pumped completely; hereafter water injection of 2 days is started, but secondary hydrate forms before gas releases from the hydrate because of rapid pressure increase caused by water injection. The volume of gas forming the secondary hydrate $\Delta V_{FH'}^{1}$ is 1060 m³ in the initial 8 h, which is calculated by the peak value of 1196 m³ minus the smallest value of 136 m³ of V_R in Fig. 5. However, volume of gas released from the hydrate deposits

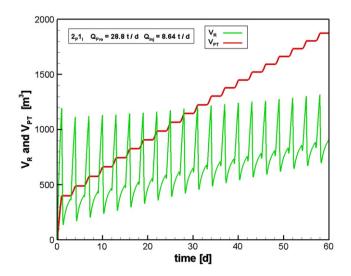


Fig.5. Volume of gas released from hydrate deposits V_R and gas production V_{PT} by huffand-puff of $1_P 2_I$.

 $\Delta V_{R''}^{1}$ in the remaining 40 h is 237 m³ (shown by the first positive slope of V_R in Fig. 5) and much less $\Delta V_{FH'}^{1}$ of 1060 m³. The amount of residual free gas ($V_R^1 = \Delta V_{R'}^1 - \Delta V_{PT}^1 = 1196-400 = 796$ m³) in the reservoir at the end of production is less than that of gas converted into secondary hydrate at the beginning of water injection process ($\Delta V_{FH'}^{1} = 1060$ m³). Thus we can conclude that the gas forming the secondary hydrate is partially from dissolved gas, $\Delta V_{H', FG}^{1} = \Delta V_{RH'}^{1} - V_{R}^{1} = 1060-796 = 264$ m³. The superscript "1" points the first time-period of the huff-and-puff operation, the subscripts *R'* and *R''* denote respectively the first and the second gas release from hydrates, *FH'* represents the first time of secondary hydrate formation in each operation period (3 days), and $\Delta V_{H', FG}^{1}$ is the volume of dissolved gas forming the secondary hydrates.

In the second period the amplitudes of all quantities are become smaller. Gas forms secondary hydrate $(\Delta V_{FH'}^2 = 14 \text{ m}^3)$ in a short time interval of <1 h at the very beginning of the second production operation (shown by the small negative peak of Q_R in Fig. 4 and small drop of V_R in Fig. 5). Volume of gas produced $\Delta V_{PT}^2 = 88 \text{ m}^3$ (the second jump of V_{PT} in Fig. 5) but released gas volume $\Delta V_{R'}^2$ is 775 m³ (the second peak of V_R in Fig. 5) during this production time interval (1 day). By comparing the peaks in Figs. 4 and 5 of this time interval, we know much of the released gas can not be produced effectively due to the low hydraulic diffusion of the porous medium. Then the operation is adjusted to water injection. Initially free gas of 943 m³ in the reservoir is converted to secondary hydrate (referred as $\Delta V_{FH''}^2$) due to pressure increase, and gas is released from hydrate due to the heat stimulation and $\Delta V_{R''}^2 = 237 \text{ m}^3$ during this injection interval. The unexpected high $\Delta V_{FH'}^2$ (=943 m³) is imaged by the second and big negative peak of Q_R in Fig. 4 and the second rapid drop from the peak point of V_R in Fig. 5. $\Delta V_{R''}^2 < \Delta V_{R'}^2$, which implies that the hydrate dissociation in the production interval due to the slight depressurization that is for producing gas efficiently is more effective than that in hot-water injection interval due to the thermal stimulation. The volume of gas converted to secondary hydrate during water injection is greater than that dissociated from the hydrate induced by heat stimulation. The low production rate but the complicated and costly operations suggest that the "huff-and-puff" operation of $(1_P 2_I)$ is not promising. Likewise, the superscript "2" denotes the second time period of the huff-and-puff, FH" represents the secondary hydrate formation in each circulatory period (3 days), and $\Delta V_{FH', FG}^1$ is the volume of dissolved gas forming the secondary hydrates.

4.2. Spatial distribution features

Because the time span of this study is short and all "phenomena" happen in a small zone that is very close to the perforated interval of the well in Fig. 2, the spatial distribution in Figs. 6 through 8 are localized in the HBL ranging from z = -60 to z = -20 m and r < 5 m. The amount of free gas released from hydrate deposits is very small in each period of huff-and-puff operation, thus the spatial distribution of S_G does not have meaningful information and will not be discussed in the flowing context.

4.2.1. Spatial distribution of T

The *T* distribution in Fig. 6 supports the low affectivity of hot water injection and also confirms the analysis above. Each separate contour plot in Fig. 6 is just at the time of ending water injection. *T* in the dissociating zone is still lower than the background temperature and has no any clear "abnormal" change revealing the heat efficiency. This result is attributed to the low water injection rate and low heat transferring in the silt clay sediment. Furthermore, the heat reached the dissociation front has been consumed by dissociating hydrate.

4.2.2. Spatial distributions of S_H

Fig. 7 shows the evolution of the S_H distribution over time in the HBL near the wellbore (r < 5 m). The characteristics of S_H distribution have never been seen before. These include (i) hydrate dissociation proceeding slowly and having a nearly "well-regulated" dissociation front, (ii) the dissociated zone performing like a growing trapezoid during the evolution, (iii) and secondary hydrate forming at the dissociation front and being more pronounced along the *z*-axis direction and with time. The unique-looking features of (i) and (ii) are because of the small dissociated scale and heat-induced dissociation that may have same dissociation rates during the "equilibrium dissociation" simulation. Secondary hydrate is caused by pressure increase during the water injection interval; its spatial evolution indicates the general loci of injected water flow as well as transferring of pore pressure, and the temporal growth implies that the heat effect on the hydrate dissociation becomes weaker because heat transferring through conduction is more and more difficult to reach the extending dissociation front but the pressure can be elevated obviously by injecting water.

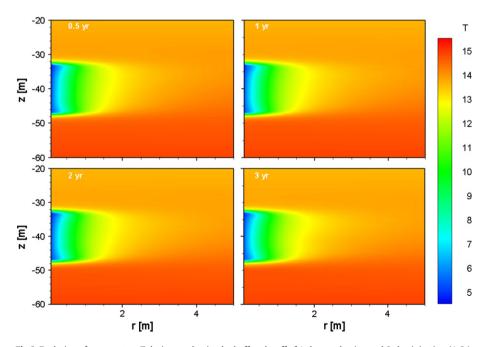


Fig.6. Evolution of temperature T during production by huff-and-puff of 1-day production and 2-day injection (1_P2₁).

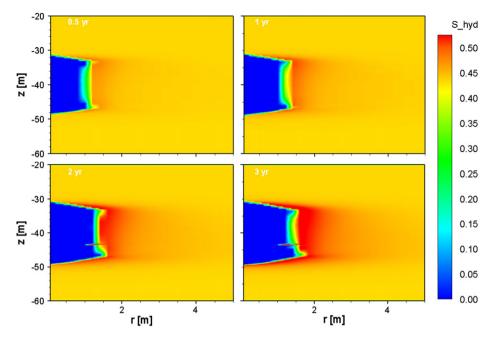


Fig.7. Evolution of hydrate saturation S_H during production by huff-and-puff of 1-day production and 2-day injection (1_P2_I).

4.2.3. Spatial distributions of X_S

The distribution of the salt concentration in Fig. 8 (shown as X_inh) illustrates the dilution effect of dissociation on salinity. Water salinity is reduced upon hydrate dissociation. The salinity reduced zone in Fig. 8 is wider than the hydrate dissociated zone in Fig. 7. A narrow blue zone indicating the most reduced salinity around the well is because of continuing removal (through production), dilution of injected water and dissociated water from hydrate, and drainage of the native saline water. The less reduced salinity (yellow zone) in the secondary hydrate zone (Fig. 7) is due to the desalination of hydrate formation. The yellow zone is pronounced along the *z*-axis direction as the operation time goes, which is consistent with the secondary hydrate that becomes more and more pronounced.

4.3. Huff-and-puff adjustment

The performance of gas production operated by huff-and-puff operation of $1_P 2_I$ discussed immediately above does not seem to be economical and of interest to industry. To further seek the effect of intermittent hot water injection on production performance, we investigate the gas production (Q_{PT} and V_{PT} , the absolute criterion) and production efficiency (R_{GW} , the relative criterion) by adjusting the length of the time interval for water injection and the mass rates of production and injection, but keep the mass ratio of production to water injection ($Q_{PT} \cdot \Delta t_{PT} / Q_{Inj} \cdot \Delta t_{Inj}$) constant. The temperature of injected water is still 80 °C. In Figs. 9 through 11, $1_P 1_I$ represents an operation of 1-day production (i.e. $\Delta t_{PTO} = 1$ day) and 1-day injection

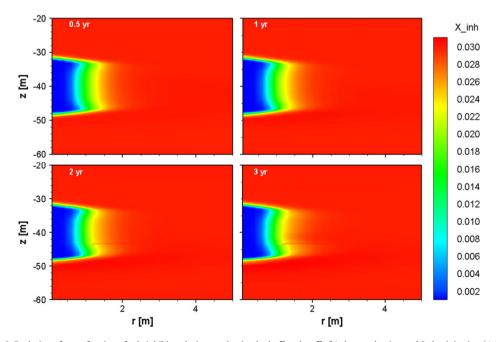


Fig.8. Evolution of mass fraction of salt-inhibitor during production by huff-and-puff of 1-day production and 2-day injection (1_P2₁).

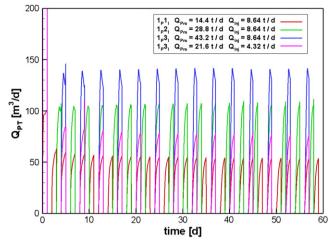


Fig.9. Evolution of gas production rate Q_{PT} at the well by huff-and-puff.

(i.e. $\Delta t_{Inj} = 1$ day), $1_P 2_1$ is a scheme of 1-day production and 2-day injection, and $1_P 3_1$ denotes an scheme of 1-day production and 3-day injection; Q_{Pro} is fluids production rate and Q_{Ini} is hot water injection rate.

Fig. 9 shows effect of different huff-and-puff operations on total gas production rate Q_{PT} at the well. Q_{PT} suggests that 1_P3_1 is the best scheme and 1_P1_1 is the worst one. The average peak value of Q_{PT} in the operation of 1_P3_1 is approximate to 140 m³/d but only 50 m³/d in 1_P1_1 operation. The effect sequence of all operations in this study is 1_P3_1 , 1_P2_1 , and 1_P1_1 , which suggests that for a fixed production time interval Q_{PT} can be increased by prolonging water injection time Δt_{Inj} . However, for a same operation period of 1_P3_1 , the higher Q_{PT} (=43.2 t/d) and higher Q_{Inj} (=8.64 t/d) have positive effect on Q_{PT} . The ratio of Q_{PT} (=140/75 = 1.87) generally has the same value with that of the production rates (=43.2/21.6=2) or injection rate (=8.64/4.32=2). The higher Q_{PT} may be attributed to the higher gas release rate Q_R induced by heat stimulation.

This behavior of Shenhu hydrate deposits in the SCS is also illustrated by the evolution of the total cumulative volume of produced gas V_{PT} and gas released from hydrate V_R . Fig. 10 shows effect of the huff-and-puff operations on V_{PT} and V_R . V_{PT} is the highest in the operation of 1_P3_1 ($Q_{Pro} = 43.2 \text{ t/d}$ and $Q_{Inj} = 8.64 \text{ t/d}$) but depressed as the injection time is reduced. Thus the sequential order of V_{PT} in the huff-and-puff operations is 1_P3_1 ($Q_{Pro} = 43.2 \text{ t/d}$ and $Q_{Inj} = 8.64 \text{ t/d}$), 1_P2_1 , 1_P1_1 , and 1_P3_1 ($Q_{Pro} = 21.6 \text{ t/d}$ and $Q_{Inj} = 4.32 \text{ t/d}$). The maximum gas

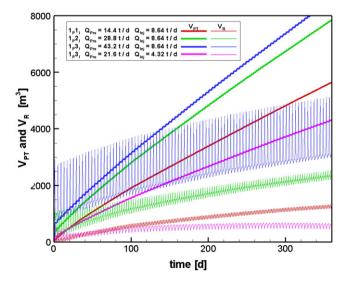


Fig.10. Evolution of cumulative volume of CH_4 produced at the well Q_{PT} and cumulative volume CH_4 released from gas hydrates Q_R in the operations of huff-and-puff.

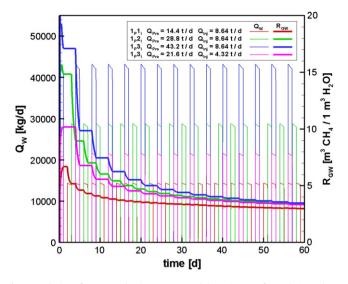


Fig.11. Evolution of water production rate Qw and the ratio R_{GW} of cumulative volume of CH₄ production V_{PT} to cumulative volume of water production M_{W} .

production in 1 year (360 days) is ~8000 m³ in the scheme of 1_P3_1 ($Q_{Pro} = 43.2$ t/d and $Q_{Inj} = 8.64$ t/d) and much less than that produced by depressurization. The sequential order of V_R is the same with that of V_{PT} . The fluctuation amplitudes of V_R are related to mass of hot water injected. More water is injected and more hydrate is dissociated by heat from hot water. Meanwhile, injecting more water (long time injection) inevitably results in more secondary hydrate. Dissociation rate is possibly be mitigated because heat flow does not effectively reach the dissociation front if the injection rate of hot water is low. That's why the volume of dissociated gas under the scheme of 1_P1_1 is greater than that in the operation of 1P31 ($Q_{Pro} = 21.6$ t/d and $Q_{Inj} = 4.32$ t/d). As a whole, V_R is much lower than V_{PT} and thus much gas produced is from the dissolved gas sink.

Fig. 11 shows the effect of the huff-and-puff operations on water production rate Q_W and gas production efficiency relative to water R_{GW} . Q_W is positively correlated to Q_{Pro} but R_{GW} is negative to Q_{Pro} . The maximum Q_W exceeds 42,000 kg/d (=42 ton/d ~42 m³/d) within each production interval of 1_P3_1 (Q_{Pro} =43.2 t/d and Q_{Inj} =8.64 t/d) but the average R_{GW} is ~5. That means that a total gas of 210 m³ can only be produced while water of 25.92 tons is injected in each period (4 days) of the huff-and-puff operation. By considering the huge water production and water injection versus the small gas production, we think the huff-and-puff operation to the Shenhu hydrate deposits are not of attraction.

5. Conclusions

We investigated a huff-and-puff production potential of gas hydrate deposits at drilling site of SH2 in Shenhu area on the northern continental slope of South China Sea. A vertical well was designed to produce gas from the perforated interval at middle section of the well in the HBL. We dissociate hydrate by alternately producing fluids and injecting hot water. Thus simulations indicated that injecting hot water did not seem as an effective way to dissociate hydrate by the heat stimulation. Most gas released from the hydrate was indeed mainly attributed to depressurization during production process. The volume of gas released from hydrate due to the heat injection was less than that converted to secondary hydrate because of the pressure elevation. In the first 60 days run by huff-and-puff operations of $1_P 2_I$ (1 day production and 2 day injection), the average gas production rate $\bar{Q}_{PT} = 93 \text{ m}^3/\text{d}$ and $V_{PT} = 1875 \text{ m}^3$, Q_W exceeded 28,000 kg/d (=28 t/d~28 m³/d), the volumetric ratio of gas to water production decreases and R_{GW} = 3.3 at t = 60 d. Furthermore, water of 345.6 t was injected in the 60 days. The hydrate dissociation

was very slow and the dissociation radius was ~1 m after a 3 year production by the huff-and-puff method. Increase injection time and heat injection rate were good to hydrate dissociation. The maximum V_{PT} is ~15,949 m³ after 3 years in the scheme of 1_P3_I but ~5500 m³ gas was only released from the hydrate deposits. The absolute criterion and relative criterion suggested that gas production from the hydrate deposits by the huff-and-puff method through a vertical well were not promising and of industrial potential. A desirable huff-and-puff production may need novel production schemes and advanced well designs.

Nomenclature

- Radial increment (m) Δr
- Time step size (s) Δt

Δz Vertical discretization, i.e., in the <i>z</i> -direction	(m))
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- С Specific heat (J/kg/K)
- k Intrinsic permeability (m^2)
- k_Θ Thermal conductivity (W/m/K)
- Thermal conductivity of dry porous medium (W/m/K) $k_{\Theta RD}$
- Thermal conductivity of fully saturated porous medium k_{ORW} (W/m/K)
- M_W Cumulative mass of water released into the ocean through the annular gravel pack (kg)
- Р Pressure (Pa)
- Mass rate of injected warm water at the well (kg/s) Q_{Ini}
- Mass rate of fluid withdrawal at the well (kg/s) Q_{Pro}
- Volumetric rate of CH_4 production at the well (ST m³/s) Q_{PT}
- Q_{PG} Volumetric rate of free gas produced at the well (ST m³/s)
- Volumetric rate of CH₄ release from hydrate dissociation Q_R $(ST m^3/s)$
- Q_W Mass rate of water produced at the well (kg/s)
- Coordinates (m) r,z
- Critical radius of maximum activity around the wellbore r_c (m)
- Radius of the well assembly (m) r_w
- Maximum radius of the simulation domain (m) r_{max}
- R_{GW} Cumulative gas-to-water ratio (ST m^3 CH₄/ m^3 water)
- S Phase saturation
- Time (davs) t
- Т Temperature (K or °C)
- V_{PT} Total volume of CH₄ produced at the well (ST m³)
- Cumulative volume of free CH₄ produced at the well (ST m³) V_{PG}
- Cumulative volume of CH₄ released from hydrate dissocia- V_R tion (ST m³)
- V_{C} Cumulative volume of CH₄ remaining in the reservoir $(ST m^3)$
- Χ Mass fraction (kg/kg)

Greek symbols

- λ van Genuchten exponent - Table 1
- φ Porosity

Subscripts and superscripts

- Aqueous phase Α
- HBL base R
- Capillary сар
- G Gas phase
- Η Solid hydrate phase irG Irreducible gas
- irA
- Irreducible aqueous phase Injection
- Ing
- Permeability reduction exponent Table 1 n
- OB Overburden
- Pro Production

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